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minute particles called atoms, and atoms are composed of even smaller components. These components

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1 moles of $\text{KClO}_3 = 122.55$ grams
1 moles of $\text{KCl} = 74.6$ grams
1 moles of $\text{O}_2 = 31.9988$ grams
Mass of test tube +

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MnO₂ -> 21.5403g Mass of test tube +
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Matter has two kinds of properties—physical and chemical. Freezing point is a physical property and ability to combine with oxygen is a chemical property. The melting of ice to

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water is an example of a physical change, and the burn-ing of wood is an example of a chemical change. A mixture is made of two or more substances. A

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